Comparison of the radioisotope dilution-coated charcoal method and a microbiological method (*L. leichmannii*) for measuring vitamin B₁₂ in serum

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SYNOPSIS A comparison has been made of the radioisotope dilution-coated charcoal method and a microbiological assay (with *L. leichmannii* as test organism) for determining the concentration of vitamin B₁₂ in serum. A satisfactory correlation was found between the results of the two methods. Under appropriate conditions the reproducibility of the radioisotope method compared favourably with that of the microbiological method.

The absence of sensitive and reliable chemical methods for measuring the trace amounts of vitamin B₁₂ in serum led to the development of appropriate microbiological methods, which because of their great sensitivity and usefulness, have been used almost exclusively. Attempts to develop chemical methods have been made in the last few years, and radiochemical procedures based on isotope dilution assays have been described (Barakat and Ekins, 1961; Rothenberg, 1961) but these methods have not supplanted microbiological assays which continue to be widely used.

More recently, Lau, Gottlieb, Wasserman, and Herbert (1965) have described a relatively simple method of employing the principle of radioisotope dilution and the absorptive properties of coated charcoal. It appears to be as sensitive and reliable as microbiological assays for measuring vitamin B₁₂ and offers some technical advantages for routine use. Vitamin B₁₂ is admirably suited to this type of assay since it can be obtained in a radioactive form (⁵⁷CoB₁₂), it has a specific binder (intrinsic factor) which is easily available and the free but not the bound form can be removed by coated charcoal.

In this paper we report our experience with the method of Lau *et al.* (1965) in measuring the concentration of vitamin B₁₂ in serum and compare the results with those obtained by a standard microbiological procedure (*L. leichmannii*).

METHODS

OUTLINE AND PRINCIPLES OF THE RADIOISOTOPE METHOD Unlabelled vitamin B₁₂ is released from serum and dilutes the specific activity of a known quantity of added ⁵⁷CoB₁₂. A representative sample of the mixture of unlabelled and radioactive B₁₂ is then bound to intrinsic factor which, having a B₁₂-binding capacity less than the amount of added ⁵⁷CoB₁₂, becomes saturated with the B₁₂ mixture. Unbound B₁₂ is removed by absorption onto protein-coated charcoal. The degree of dilution of the radioactive B₁₂ is proportional to the amount of unlabelled B₁₂ in the mixture and this will be reflected in the relative amounts bound to intrinsic factor. The concentration of B₁₂ in the serum can be calculated by comparing the quantity of radioactive B₁₂ bound to intrinsic factor in the presence of the test serum with that in a control containing only intrinsic factor and an identical amount of ⁵⁷CoB₁₂.

MICROBIOLOGICAL ASSAY Microbiological assay of serum vitamin B₁₂ level was carried out using *Lactobacillus leichmannii* as the test organism according to the method of Rosenthal and Sarett (1952). The organism (*L. leichmannii*, N.C.I.B. 8117) was obtained from the Torrey Research Station, Aberdeen, Scotland.

ISOTOPE DILUTION METHOD The isotope dilution method was carried out as described by Lau *et al.* (1965) with slight modifications.

1 PREPARATION OF TEST SERUM AND CONTROLS Into a 20 ml. screw-capped glass bottle (a Universal container) were pipetted 3.0 ml. serum, 9.0 ml. saline, and 3.0 ml.

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N/4 hydrochloric acid. The mixture was shaken and then placed in a boiling water bath for 15 minutes (with the screw-cap in position to avoid volume loss by evaporation). After the bottle had cooled at room temperature, 3 ml. ⁵⁷CoB₁₂ was added and the contents shaken. For serum supernatant controls, 3 ml. of the heated mixture was transferred to each of two 10 ml. polystyrene tubes, each containing 1.5 ml. saline (tubes 1 and 2). To the remaining mixture in the Universal container were then added 2 ml. working solution of intrinsic factor and 4 ml. of saline and the contents well mixed. From this mixture 4.5 ml. volumes were transferred to each of three 10 ml. polystyrene tubes (tubes 3, 4, and 5). To each of the five polystyrene tubes was added 2 ml. of coated charcoal suspension. These tubes were well mixed, centrifuged at 2,000 r.p.m. for 10 minutes and then 5 ml. of supernatant from each was transferred to counting tubes. (Volumes of exactly 5 ml. were required for the particular counting equipment.)

The intrinsic factor controls were prepared as follows: the initial mixture contained 12 ml. saline and 3 ml. N/4 HCl. Since no serum was present, there was no need for heating, and 3 ml. ⁵⁷CoB₁₂ was added immediately. For the intrinsic factor supernatant controls, 3 ml. of the mixture was transferred to each of two 10 ml. polystyrene tubes (tubes 6 and 7), each containing 1.5 ml. saline. To the remaining mixture were added 2 ml. working solution of intrinsic factor and 4 ml. saline; 4.5 ml. of this mixture was then transferred to each of three 10 ml. polystyrene tubes (tubes 8, 9, and 10). The subsequent treatment was identical to that for the sera.

2 COUNTING OF RADIOACTIVITY Each sample was counted for 100 seconds in a well-type scintillation counter.

3 CALCULATION OF SERUM VITAMIN B₁₂ CONCENTRATION was by means of the formula of Lau *et al.* (1965).

$$\text{pg B}_{12} \text{ per ml. serum} = C \times \left(\frac{IF}{S} - 1 \right)$$

where C = B₁₂ concentration in pg./ml. of ⁵⁷CoB₁₂ standard

S = net counts per 100 seconds of test serum.
= average of tubes 3, 4, and 5 — average of tubes 1 and 2.

IF = net counts per 100 seconds of intrinsic factor control
= average of tubes 8, 9, and 10 — average of tubes 6 and 7.

4 PREPARATION OF INTRINSIC FACTOR A hog intrinsic factor concentrate (batch no. 48743/131) supplied by Lederle Laboratories was used. (This preparation was also being used routinely for Schilling tests.) A stock solution of 100 mg./100 ml. was prepared and after filtration through Whatman no. 1 filter paper its B₁₂-binding capacity was measured by the method of Gottlieb, Lau, Wasserman, and Herbert (1965). This stock solution was stable for at least six months at -20°C., but the working solution was prepared freshly for each batch of assays because it was found that the diluted solution lost

25 to 30% of its activity after a week at 4°C. It was also found best for the assay to keep the B₁₂-binding capacity of the working solution as close to 800 pg./ml. as possible. The working solution was made by adding 1 ml. of the stock solution to 89 ml. of deionized water.

5 ALBUMIN - COATED CHARCOAL Activated charcoal powder (B.D.H.) and 30% bovine albumin (Armour) were used. Ten g. of charcoal and 6.7 ml. of albumin were mixed and made up to 400 ml. with deionized water. This was stored at 4°C. and remained active for at least a month. This was sufficient for the assay of 40 sera but was also used for intrinsic factor antibody tests and in practice a new batch was prepared every fortnight. The suspension was well shaken before use.

6 ⁵⁷CO VITAMIN B₁₂ (⁵⁷COB₁₂) Two batches, one 11 μC./μg. and the other 15 μC./μg., were used. These were obtained from the Radiochemical Centre, Amersham, Bucks. Standardization of each batch was by reverse isotope dilution against the cyanocobalamin standard (Lau *et al.*, 1965).

7 VITAMIN B₁₂ (CYANOCOBALAMIN) STANDARD 1,000 pg./ml. was prepared from 100 μg./ml., Cobalin (Paines and Byrne) brand of cyanocobalamin B.P.

8 SOURCE OF SERA Seventy-five sera were assayed. Some sera were subsequently re-assayed to determine the reproducibility of the methods. Altogether 106 assays by isotope dilution and 103 by the microbiological method were carried out. Twelve sera were obtained from patients with untreated pernicious anaemia. The remainder were obtained from patients with a variety of other diseases and from normal subjects. The sera were stored for up to two months at -20°C. before assay.

RESULTS

The results of the assays on 75 different sera are

FIG. 1. Comparison of individual serum vitamin B₁₂ results (75 sera) obtained by radioisotope dilution and microbiological (*L. leichmannii*) methods.

TABLE I

SUMMARY AND COMPARISON OF SERUM VITAMIN B₁₂ RESULTS OBTAINED BY RADIOISOTOPE DILUTION AND MICROBIOLOGICAL (*L. leichmannii*) METHODS

Serum B ₁₂ Level (pg./ml.)	Number of Sera	
	Co ⁵⁷ B ₁₂ Method	<i>L. leichmannii</i> Method
<150	13	12
150-200	11	6
>200	51	57
Total number of sera	75	75

shown in Fig. 1 and Table I. Comparison of the values obtained by the radioisotope and microbiological methods yielded a correlation coefficient of 0.89. It will be seen that over the range of vitamin B₁₂ concentration examined there was a satisfactory agreement between the two methods but that the isotope dilution method tended to give lower values when the vitamin B₁₂ concentration was less than 200 pg./ml.

The reproducibility of each method was assessed by replicate assays on one serum (sample d) set up in a single batch, and also by repeated assays over a period of five months on each of three different sera (sample a, b, and c). These results are shown in Table II from which it can be seen that for practical purposes the reproducibility of the assays by the radioisotope method was satisfactory, and that in this respect it compared favourably with the microbiological method. To obtain this degree of reproducibility with the radioisotope method it was found necessary to use larger volumes of serum and

reagents than those recommended by Lau *et al.* (1965).

DISCUSSION

The results of this study substantiate the claim of Lau *et al.* (1965) that their radioisotope method can be used to determine the amount of vitamin B₁₂ in serum and that the values correlate reasonably well with those obtained by microbiological assay. Our study differs from that of Lau *et al.* (1965) in that we have used *Lactobacillus leichmannii* and they have used *Euglena gracilis* as the test organism for the microbiological assay. The radioisotope method would therefore seem to correlate favourably with either of the standard microbiological methods. Like Lau *et al.* (1965) we have found that the radioisotope method tends to give lower values than the microbiological method on sera containing reduced amounts of vitamin B₁₂. This might mean that vitamin B₁₂-deficient sera can be identified with greater certainty by the radioisotope method but many more sera would have to be assayed to know whether this is a consistent trend. What appears certain is that the radioisotope method is at least as reliable as the microbiological method in picking out B₁₂-deficient sera.

In order to obtain better reproducibility of results with the radioisotope method we have used larger volumes of serum and reagents than those used by Lau *et al.* (1965). It was found that larger volumes could be dispensed more easily and accurately and

TABLE II

REPRODUCIBILITY OF SERUM VITAMIN B₁₂ RESULTS OBTAINED BY THE RADIOISOTOPE DILUTION METHOD AND BY MICROBIOLOGICAL ASSAY

Method	Times of Assay	Serum	Number of Replicate Assays	Serum Vitamin B ₁₂ Concentration (pg./ml.)				
				Individual Values				Mean
Isotope dilution	Over a period of 5 months	a	4	10	0	50	80	35
		b	6	280 400	350 520	370	360	380
		c	7	1,080 760	850 760	890 890	990	890
	In a single batch	d	9	510 530 500	500 530	520 530	540 550	525
		a	5	30 40	60	10	<25	30
		b	13	400 450 390 435	500 480 430	420 450 410	440 410 425	430
<i>L. leichmannii</i>	Over a period of 5 months	c	3	860	660	750		760
	In a single batch	d	4	480	480	480	420	465

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that this modification greatly improved the reproducibility and accuracy of the method in our hands.

Although results can be obtained in a matter of hours with the radioisotope method compared with at least three days for microbiological methods, there is seldom a pressing need to know the serum B₁₂ level quickly. The main advantage of the relative rapidity of the assay is the saving of technicians' time: a batch of sera assayed radioisotopically takes less time than the same number assayed microbiologically. A potential advantage of the radioisotope method is that it appears to lend itself more easily to automation which might allow large-scale assays to be carried out, perhaps in regional centres.

In terms of material the radioisotope method costs about six times as much as the microbiological method, *i.e.*, about 2s. as opposed to 4d. per serum sample. The greater expense is largely due to the cost of the ⁵⁷CoB₁₂. Furthermore, the isotope method requires relatively specialized counting equipment and unless an automatic counter is incorporated (which increases the cost of the equipment) the counting of large numbers of samples can be tedious and time consuming. Microbiological methods can be set up in almost any laboratory since the materials are easily available and no specialized equipment is necessary. No particular safety precautions are required for the microbiological procedures whereas there is the potential hazard of working with radioactive materials in the isotope methods.

Almost all laboratories carrying out vitamin B₁₂ assays use one or other of the standard microbiological methods. Once established these methods work satisfactorily and it remains to be seen whether

a familiar and reliable microbiological method will be readily abandoned for the radioisotope procedure.

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ADDENDUM

Since this paper was accepted for publication, it has been found that occasionally and unpredictably in some assay batches, the net counts per 100 seconds of the intrinsic factor control have been a little lower than expected and so the serum B₁₂ values for the sera assayed in that batch have also been lower than expected. This may be due to the fact that while there are serum proteins in the serum tubes which may protect the intrinsic factor against damage by acid, there are none in the intrinsic factor control tubes.

Now in every batch, a serum sample known to contain < 25 µg./ml. of vitamin B₁₂ is assayed, and if the net counts per 100 seconds of the serum are higher than the net counts per 100 seconds of intrinsic factor control, the serum value is the one used for the value IF in the equation for the calculation of serum B₁₂ concentration.

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