New syringe type of microtitre apparatus for virus and other serological tests

J. G. ALEXANDER AND J. TEAL From the Virus Laboratory, Castle Hill Hospital, Cottingham, Yorkshire

Until now the standard micromethod for viral serological tests has been that of Takátsy (1955) or one of its modifications. The method employed in this laboratory is a miniature version of the standard macromethod (Bradstreet and Taylor, 1962), using 0·1 ml volumes. Their method has been compared with the version to be described. Diluent, complement, antigen, and red cell suspension were added by means of a Hamilton repeating dispenser in 20 microlitre volumes, the apparatus giving 50 such volumes per fill. A Marburg syringe with a disposable yellow tip was used to add sera and to make doubling dilutions. These items of apparatus are obtainable in Britain and are cheaper than the apparatus described by Takátsy (1955) or its modifications.

Disposible microtitre plastic plates with U-shaped wells were used. The yellow tip on the Marburg syringe was rinsed exactly as a standard automatic syringe is rinsed, but a 'blow-out' mechanism was incorporated in the Marburg syringe.

These items of apparatus were tried out in the routine virus and other serological tests carried out in this laboratory. The antigens routinely used were influenza A, B, and C, parainfluenza I, psittacosis LGV, respiratory syncytial, mumps S, mumps V, Herpes simplex and measles viruses, adenovirus, Rickettsia burnetii, Mycoplasma pneumoniae, Leptospira icterohaemorrhagiae, campylobacter, and biflexa.

In 93 screening tests at 1/10 and 197 titrations (1/10 to 1/320) the results were shown to agree with those obtained using the macromethod. No differences of more than one dilution were observed in the titrations, and, where such differences occurred, they were reproducible and made no difference to the degree of rise in titre. There were no false positive or false negative results in the screen tests.

This microtechnique has now been adopted exclusively. It may not be quicker than the Takátsy method, but reagents are delivered as accurate volumes and not as drops, and the method is more convenient and more easily carried out by junior staff.

REFERENCES


Received for publication 12 November 1968.

Letters to the Editor

Having followed with interest the correspondence concerning radioisotopic dilution techniques for vitamin B₁₂ assay we would like to describe a modification to one of the techniques which improves its accuracy and reproducibility.

Using the technique of Lau, Gottlieb, Wasserman, and Herbert (1965) for the assay of vitamin B₁₂ activity in serum, in which the intrinsic factor control is prepared using saline, we observed, as did Raven, Walker, and Barkhan (1966) and subsequently Downer and Oliver (1968), that occasionally the lower vitamin B₁₂ 'test' net count rates were higher than the intrinsic factor control net count rates. This anomaly will result in the theoretical calculation of a negative serum vitamin B₁₂ activity as well as a false lowering of vitamin B₁₂ activities generally, since the intrinsic factor control net count rate represents the maximal count rate of isotopic labelled vitamin B₁₂ before dilution with 'cold' vitamin B₁₂. Falsely low values are particularly undesirable when vitamin B₁₂ activities in the low normal range are reduced, somewhat haphazardly, to pathological values. At least two attempts to overcome this difficulty have been described. Raven et al. (1966), and subsequently Downer and Oliver (1968), substituted the lowest vitamin B₁₂ 'test' net count rate in place of the intrinsic factor control net count rate in the calculation of the serum vitamin B₁₂ activities in any batch of assays in which this difficulty was encountered. This occasional and unpredictable substitution must produce erratic and unreproducible values. More recently, Raven et al. (1967) reported experiences using a 1 : 10 dilution of a vitamin B₁₂-deficient serum in the preparation of the intrinsic factor control. However, we feel that this is not the complete answer, since intrinsic factor has been shown to exhibit varying binding properties of vitamin B₁₂ in different solutions, especially in solutions of varying concentrations of serum extracts (Rothenberg, 1968).

While attempting to find a suitable serum or protein medium for the preparation of the intrinsic factor control we accidentally observed that serum to which ascorbic acid had been added for the purpose of 'folute assay' gave consistently low vitamin B₁₂ values by the technique of Lau et al. (1965). This led to the adoption, after further investigation, of an ascorbic-acid-saturated serum as the medium used to prepare the intrinsic factor control. Using this medium we have found, in a series of over 1,500 investigations, no serum that has given a higher net count rate than the intrinsic factor control net count rate and the difficulty outlined above has therefore never been experienced. Further, we find that when this treated serum is used to prepare the intrinsic factor control it gives consistent net count rates from week to week, and any newly prepared batch of treated serum also gives net count rates that correspond accurately with those obtained from the previous intrinsic factor control.

A further important consequence of this effect of
ascorbic acid is the error which will result from accepting a 'folate serum', preserved by ascorbic acid, for vitamin B₁₂ assay. Using radioisotopic dilution techniques dependent on intrinsic factor binding such a serum will exhibit a misleadingly low vitamin B₁₂ activity in a clinical situation where a low level might be anticipated.

Having noted the effect of ascorbic acid in vitro we considered the possibility of an effect in vivo and now have evidence that very high oral doses of ascorbic acid can produce a lowering of vitamin B₁₂ activity as measured by this technique.

We hope to publish the full details of our modifications to Lau's original technique and the results of our further studies.

E. GALLIE and J. ROBSON
Stirlingshire Area Laboratory Service, Falkirk Royal Infirmary, Falkirk

REFERENCES

NORMAL VALUES FOR INDIVIDUAL PLASMA COBALAMINS

In this letter we wish to make a preliminary communication of normal values for individual plasma cobalamins, and to report values obtained in two cases of untreated pernicious anaemia. It has been known for some years that the total serum (or plasma) vitamin B₁₂ is made up of several compounds, of which methylcobalamin is usually predominant (Lindstrand and Ståhlberg, 1963). Quantitative values for the individual cobalamins have not, however, yet been published.

Heparinized blood samples were taken from 20 healthy people, 17 of whom were smokers. Plasma cobalamins were separated and located by thin-layer chromatography and bioautography as described by Linnell, MacKenzie, Wilson, and Matthews (1969). The proportion of each fraction was estimated by photometric scanning of stained growth areas and by comparison with standard cobalamin solutions treated similarly to the plasma extracts (Linnell, MacKenzie, and Matthews, 1969). Total plasma B₁₂ was estimated by radioisotopic assay (Matthews, Gunasegaram, and Linnell, 1967).

In all normal samples, the predominant component was methylcobalamin. The second major component (recorded as hydroxocobalamin but probably also containing some deoxyadenosyl coenzyme B₁₂, averaged about one quarter of the total B₁₂; the range for this component was extremely wide. Six samples showed a small proportion of cyanocobalamin; in most of these it was negligible, but in one individual it amounted to about 10% of the total B₁₂. The results suggest a tendency to slight overestimation of the hydroxocobalamin and cyanocobalamin growth areas when chromatograms are assessed visually (Linnell et al., 1969).

In the cases of pernicious anaemia, the pattern of distribution of the two major components was abnormal, the ratio of methylcobalamin to hydroxocobalamin, normally greater than 1:1, being very much reduced. In both cases, hydroxocobalamin, though low, was still within normal limits. The significance of this abnormality is not yet clear.

We are grateful to Dr John Wilson and Dr A. V. Hoffbrand for providing many of the blood samples. The work was supported by a grant from the Wellcome Trust.

J. C. LINNELL,
H. M. MACKENZIE,
D. M. MATTHEWS,
Department of Chemical Pathology,
Westminster Medical School,
London, SW1.

REFERENCES

VALUES FOR INDIVIDUAL PLASMA COBALAMINS IN NORMAL SUBJECTS AND CASES OF UNTREATED PERNICIOUS ANAEMIA

<table>
<thead>
<tr>
<th>Total B₁₂ (μg/ml)</th>
<th>Me-B₁₂ (μg/ml)</th>
<th>CN-B₁₂ (μg/ml)</th>
<th>O₂B₁₂ (μg/ml)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Twenty normal subjects (mean, SE, range)</td>
<td>544 ± 45 (250-1085)</td>
<td>398 ± 33 (158-635)</td>
<td>5.5 ± 2.8 (0-45)</td>
</tr>
<tr>
<td>Individual values for two cases of pernicious anaemia</td>
<td>30</td>
<td>7</td>
<td>&lt;5</td>
</tr>
<tr>
<td></td>
<td>65</td>
<td>20</td>
<td>45</td>
</tr>
</tbody>
</table>