Extraction of serum vitamin B₁₂ for radio-isotopic and Lactobacillus leichmannii assay

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SYNOPSIS The protein precipitates discarded during the extraction process of the Lactobacillus leichmannii vitamin B₁₂ assay have been shown to contain significant amounts of vitamin B₁₂. This loss of vitamin B₁₂ provides a satisfactory explanation for many of the discrepancies between the serum vitamin B₁₂ values obtained by the L. leichmannii method and the radio-isotopic method of Raven et al (1969). It is possible to produce lower results by the method of Raven et al (1969) by incorporating into that method the L. leichmannii extraction process; it is also possible to produce higher results by the L. leichmannii method using a papain extraction process.

Over the last nine years, a radio-isotopic assay of vitamin B₁₂ in serum has been found to give higher results for many sera than the Lactobacillus leichmannii microbiological assay. For some sera from post-gastrectomy patients and from patients with folate deficiency of many different causes the discrepancies in results have been especially marked and often of clinical importance (Raven et al, 1969; Raven et al, 1972). This paper gives the results of investigations into the reasons for the differences between the results obtained by the radio-isotopic and microbiological assays.

Material and Methods

RADIO-ISOTOPIC STUDIES

The radio-isotopic assay was that of Lau et al (1965) modified by Raven et al (1969). Individual backgrounds were used for each serum sample, and the normal range for this method is 200-1000 pg/ml. In this method, serum vitamin B₁₂ is extracted from its binding proteins by the heating of serum in the presence of cyanide and N/4 HCl. A turbid solution is obtained and no precipitation of protein occurs (fig 1).

For the assay of the serum extracts and protein precipitates obtained from the extraction process used in the L. leichmannii assay, the radio-isotopic assay was modified as follows:

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An appropriate intrinsic factor control is prepared similarly from an L. leichmannii serum extract of either a serum containing less than 10 pg/ml of vitamin B₁₂ (by radio-isotopic assay) or a 1 in 20 dilution of vitamin B₁₂-deficient serum. The vitamin B₁₂ value derived by calculation is multiplied by 10/8 to give the serum vitamin B₁₂ level.

L. leichmannii Protein Deposits
The protein deposit obtained from 1 ml serum by means of the L. leichmannii extraction process is placed in a 20 ml universal container and washed thoroughly with three changes of distilled water. After the last wash, the supernatant fluid is removed, and the volume of the deposit is made up to 10 ml with 2 ml N/4 HCl and the required volume of 0·1% (w/v) NaCN solution. With the screw cap in position, the container is then autoclaved at 115°C for 15 minutes and allowed to cool. The extract is then treated as in the radio-isotopic assay above. An appropriate intrinsic factor control is prepared in the same way from the protein deposit obtained from a serum shown by radio-isotopic assay to have a B₁₂ content of less than 10 pg/ml.

L. leichmannii Studies
The L. leichmannii assay was that of Rosenthal and Sarett (1952) modified by Spray (1955) and Chanarin (1969). It is the method used in the previous studies of Raven et al. (1969; 1972) and is described in detail by Chanarin (1969). Its normal range is 150-850 pg/ml, although occasionally sera from patients with untreated pernicious anaemia have vitamin B₁₂ values of between 150 and 200 pg/ml (Ardeman et al., 1966; Raven et al., 1972). In this method, serum vitamin B₁₂ is extracted from its binding proteins by a deproteinization technique involving the heating of serum in the presence of cyanide and acetate buffer (pH 4·5). Only the clear supernatant obtained is used in the L. leichmannii assay (fig 1). The protein precipitate is discarded before assay. For some studies the L. leichmannii method was modified to incorporate a papain extraction process as follows:

Preparation of Papain Solution
Ten grams of papain powder (Merck) is ground in a mortar with 100 ml Sorensen’s phosphate buffer (0·066M, pH 5·4), and the mixture is removed and centrifuged for 10 minutes. Ten ml of 0·5M cysteine HCl solution is added to the supernatant to activate the papain, and the mixture is made up to 200 ml with phosphate buffer and incubated at 37°C for one hour. The 5% papain solution thus obtained is dispensed into 25 ml aliquots and stored for up to six months at −20°C.

Papain—L. leichmannii extraction process
Into a 20 ml universal container are placed 1 ml serum, 7 ml water, 0·5 ml 0·1% NaCN solution, and 1 ml 5% papain solution. The container is then heated in a 60°C water bath for one hour and, after the addition of 0·25 ml 0·4M acetate buffer (pH 4·5) autoclaved at 115°C for 10 minutes. After cooling, 0·25 ml phosphate buffer (M/5 K₂HPO₄) is added and the universal container is placed in a 4°C refrigerator overnight to allow the fine protein particles to precipitate, leaving a clear supernatant (fig 1). (The protein particles are so fine that centrifugation often produces a turbid supernatant.) For the L. leichmannii assay, this supernatant is then treated in exactly the same way as the ordinary L. leichmannii serum extract. The standard curve for the papain-L. leichmannii assay is set up in the same way as for the standard L. leichmannii assay.

Results

Radio-Isotopic Studies

Assay of L. leichmannii Serum Extracts
The vitamin B₁₂ levels of 331 miscellaneous sera were determined by the radio-isotopic and L. leichmannii methods and by the radio-isotopic method modified to incorporate the L. leichmannii extraction process. The radio-isotopic method produced generally higher values than the L. leichmannii method (fig 2); there was closer agreement between the L. leichmannii method and the radio-isotopic method using the L. leichmannii extraction process.
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Fig 3. Comparison of vitamin B₁₂ values obtained for 331 sera by the L. leichmannii method and the radio-isotopic method of Raven et al. (1969), modified to incorporate the L. leichmannii extraction process.

Similar conclusions may be reached about the results obtained by the three different assays for 110 sera from post-gastrectomy patients, 45 folate deficient sera, and 40 sera from patients with vitamin B₁₂ deficiency due to untreated pernicious anaemia (figs 4 to 6).

Fig 4. Vitamin B₁₂ values obtained for 110 sera from post-gastrectomy patients by the radio-isotopic method, the radio-isotopic method modified to incorporate the L. leichmannii extraction process, and the L. leichmannii method. In this and subsequent figures, the horizontal lines represent mean values.

Assay of L. leichmannii protein deposits

The amount of vitamin B₁₂ detected in the protein deposits obtained from 331 miscellaneous sera by means of the L. leichmannii extraction process ranged from 0 to 365 pg per ml of whole serum. The mean value was 65 pg (fig 7). The values for 48 vitamin B₁₂ deficient sera, 48 folate deficient sera, 138 sera from post-gastrectomy patients, and 70 normal sera are shown in fig 8, the highest values being given by the folate deficient and post-gastrectomy groups of sera.

The agreement between radio-isotopic and L. leichmannii methods (fig 2) is improved when to the L. leichmannii value is added the amount of vitamin B₁₂ detected in the L. leichmannii protein deposit (fig 9).
Fig 7  Distribution of amounts of vitamin B\textsubscript{12} detected by the radio-isotopic method of Raven et al (1969) in the protein deposits obtained from 331 miscellaneous sera. The mean value is 65 pg/ml of whole serum.

Fig 8  Amounts of vitamin B\textsubscript{12} detected by the radio-isotopic method of Raven et al (1969) in the protein deposits (from 1 ml of whole serum) obtained from 48 B\textsubscript{12} deficient sera, 48 folate deficient sera, 138 post-gastrectomy sera, and 70 normal sera.

L. leichmannii studies

Tests for the Presence of L. leichmannii Inhibitor
(a) Twenty sera shown to have considerably higher serum vitamin B\textsubscript{12} values by the radio-isotopic assay than the L. leichmannii assay were mixed with equal parts of a serum whose vitamin B\textsubscript{12} value was found to be 500 pg/ml by both radio-isotopic and L. leichmannii assays. The serum mixtures were then assayed by the L. leichmannii method. For each mixture the vitamin B\textsubscript{12} values obtained were the same as those expected by calculation, thus excluding the presence of some L. leichmannii inhibitor in the sera tested.

(b) Replicate vitamin B\textsubscript{12} assays were carried out by the L. leichmannii method for 397 sera, first using the usual volume of serum extract and, secondly, twice this volume (with the results divided by two to obtain the serum vitamin B\textsubscript{12} value). The results obtained (fig 10) show the double-volume values to be a little lower than the single-volume values. However, the differences are of little importance and

Fig 9  Comparison of vitamin B\textsubscript{12} values obtained for 331 sera by the radio-isotopic method of Raven et al (1969) and by the L. leichmannii method, when to each L. leichmannii value has been added the amount of vitamin B\textsubscript{12} detected in the L. leichmannii protein deposit by the radio-isotopic method.

Fig 10  Comparison of vitamin B\textsubscript{12} values obtained for 397 sera by the L. leichmannii method in replicate assays, one assay using a single volume of serum extract and the other a double volume.
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vitamin B₁₂ solutions and vitamin B₁₂ deficient serum, the mixtures for one curve being treated by means of the standard L. leichmannii extraction process and for the other by the papain modification. The two curves thus obtained were similar on each occasion.

Discussion

Quite naturally, the early radio-isotopic assays of vitamin B₁₂ in serum were expected to produce results as close as possible to the conventional microbiological assays (L. leichmannii and Euglena gracilis). The method of Lau et al (1965) was reported to give a good correlation with the E. gracilis assay but it soon became apparent that there was a defect in this radio-isotopic method, tending to produce falsely low serum vitamin B₁₂ values or even negative values for vitamin B₁₂ deficient sera (Raven et al, 1966, 1968, 1969; Hillman et al, 1969); many modifications have been suggested to overcome this defect which was due to the absence of serum proteins from the standards (intrinsic factor controls) (Raven, 1970).

Experience since 1966 with a modification of the method of Lau et al (1965) has shown that in general the serum vitamin B₁₂ values obtained by the radio-isotopic assay are higher than the L. leichmannii microbiological method (Raven et al, 1969; 1972). The discrepancies have been particularly marked for some folate deficient sera and post-gastrectomy sera. Even the unmodified method of Lau et al (1965) has been reported to give higher vitamin B₁₂ values than the L. leichmannii method for some sera from pregnant women (Lawrence and Klipstein, 1967).

The purpose of the investigations in this paper was to find the reason for differences in the results obtained by the radio-isotopic and L. leichmannii assays. The possible explanations included:

TECHNICAL DEFECT IN THE RADIO-ISOTOPIC METHOD

To anyone with long experience with a microbiological vitamin B₁₂ assay and a belief in the accuracy of its results this would seem the likely explanation. The radio-isotopic results would then be considered falsely high. In the radio-isotopic method of Raven et al (1969), some defect in the acid extraction process, the binding of vitamin B₁₂ by intrinsic factor, the separation of free and bound B₁₂ by albumen-coated charcoal or in the calculation of results by equation could lead to consistently high serum vitamin B₁₂ values but it would be difficult to explain why some folate deficient or post-gastrectomy sera should give especially discrepant
results and why accurate values should be given for vitamin B\textsubscript{12} solutions and dilutions of sera. It has already been shown that the calculation of results by equation gives similar results in the 0-1000 pg/ml range to those obtained from a standard curve (Raven et al, 1968) and, in addition, other quite different radio-isotopic assays have produced higher serum vitamin B\textsubscript{12} values than the \textit{L. leichmannii} assay (Raven and Robson, 1974; van de Wiel et al, 1974). The acid extraction process, used in the method of Raven et al (1969) and suggested by Rothenberg (1968) as a cause of falsely high serum vitamin B\textsubscript{12} values, will be discussed later.

**DETECTION OF DIFFERENT FORMS OF VITAMIN B\textsubscript{12}**

Both the radio-isotopic method of Raven et al (1969) and the \textit{L. leichmannii} method use cyanide in their extraction processes and so both assays are detecting cyanocobalamin. There remains the possibility that some analogue of vitamin B\textsubscript{12} which does not support the growth of \textit{L. leichmannii} is also being detected by the radio-isotopic assay. It is difficult to exclude this possibility but the experience of Raven and his colleagues over the last nine years is that when the radio-isotopic assay gives a normal value and the \textit{L. leichmannii} assay a subnormal value on the same serum, a full clinical and haematological study fails to detect any evidence of vitamin B\textsubscript{12} deficiency. Also the sera from all patients with haematological evidence of vitamin B\textsubscript{12} deficiency have given subnormal vitamin B\textsubscript{12} values by the radio-isotopic method, whether the vitamin B\textsubscript{12} deficiency was due to pernicious anaemia or followed gastric surgery. Any hypothetical vitamin B\textsubscript{12} analogue cannot therefore be distorting the serum vitamin B\textsubscript{12} values so that false clinical information is being given by the radio-isotopic assay. These clinical findings, the results of extraction and papain studies described in this paper, and a comparison of \textit{L. leichmannii} and \textit{E. gracilis} assays (Raven et al, 1972), all make it unlikely that some vitamin B\textsubscript{12} analogue is responsible for the discrepancy between radio-isotopic and \textit{L. leichmannii} results.

**SERUM L. leichmannii INHIBITOR**

Such an inhibitor causing falsely low \textit{L. leichmannii} values has been postulated by Lawrence and Klipstein (1967) but no good evidence, including the experiments described in this paper and those of Green et al (1974), has ever been found except, of course, such obvious examples as sera containing antibiotics.

**TECHNICAL DEFECT IN THE L. leichmannii METHOD**

The variability of results that one might expect from a microbiological assay can be seen both from studies of the reproducibility of the \textit{L. leichmannii} method (Spray, 1955; Matthews, 1962; Raven et al, 1966; Wagstaff and Broughton, 1971) and from its comparison with the \textit{E. gracilis} assay (Raven et al, 1972) but cannot be an explanation for any systematic difference in radio-isotopic and \textit{L. leichmannii} results. One possible explanation, however, is the difference in extraction processes used by the radio-isotopic and \textit{L. leichmannii} assays to separate vitamin B\textsubscript{12} from its binding proteins. The radio-isotopic assay uses a whole serum method in which serum is heated in the presence of cyanide and N/4 HCl. A turbid solution is obtained and, because of the low pH (approximately 2-0), no protein precipitation occurs. The \textit{L. leichmannii} method uses a deproteination technique in which serum is heated in the presence of cyanide and an acetate buffer (pH 4-5). A clear supernatant is used for the vitamin B\textsubscript{12} assay and a flocculent white protein precipitate is discarded.

Ever since the introduction of cyanide into the \textit{L. leichmannii} extraction process it has been believed that no vitamin B\textsubscript{12} remains in the protein precipitate and that all the serum vitamin B\textsubscript{12} is available for assay (Matthews, 1962). Only occasionally has it been suggested that the extraction of vitamin B\textsubscript{12} by a deproteination method may lead to falsely low serum vitamin B\textsubscript{12} values (Barakat and Ekins, 1963; Carmel and Colman, 1969; Raven et al, 1969). The studies described in this paper show that often significant amounts of vitamin B\textsubscript{12} are left behind in the protein precipitates and so are discarded before \textit{L. leichmannii} assay. This loss of vitamin B\textsubscript{12} provides an explanation for the \textit{L. leichmannii} results being generally lower than those of the radio-isotopic assay. The especially discrepant results for some folate deficient or post-gastrectomy sera can be explained by the increased loss of vitamin B\textsubscript{12} in the protein precipitates formed from these groups of sera (fig 8).

Although these findings supported the belief that the difference in extraction processes was at least one of the factors responsible for the differences in radio-isotopic and \textit{L. leichmannii} results, it was felt that the evidence would be stronger if it were possible to modify the microbiological method so that it could produce higher serum vitamin B\textsubscript{12} values. It proved possible to convert the acidic serum extract used in the radio-isotopic assay into a form suitable for \textit{L. leichmannii} assay by heating it in the presence of Difco brand B\textsubscript{12} assay medium, USP. Complete protein precipitation occurred and...
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a clear extract of appropriate pH was obtained. Unfortunately, the vitamin $B_{12}$ results were similar to those obtained by standard *L. leichmannii* assay. It did prove possible to produce higher *L. leichmannii* values by adding another step to the serum extraction process; before centrifugation to remove precipitated proteins, N/4 HCl was added and the extract was heated a second time. However, for some sera the turbidity of their extracts was too great, and individual serum blanks were necessary. It was also necessary to modify the standard curve so that the pH of the standard curve and serum tubes was similar. Because of these difficulties it was decided to try a completely different form of extraction process using some proteolytic enzyme.

Enzyme modifications of microbiological assays had been used previously in the assay of vitamin $B_{12}$ in serum, milk, bile, and tissue. Gregory (1954) had shown that prior digestion with cyanide-activated papain was necessary before the vitamin $B_{12}$ content of human and cow's milk could be measured by the *L. leichmannii* assay, and both Shenoy *et al* (1957) and Hall and Allen (1964) had shown that higher serum vitamin $B_{12}$ values could be obtained by the *E. gracilis* method if a papain extraction process were used. The papain-*L. leichmannii* method described in this paper proved simple to use and produced higher serum vitamin $B_{12}$ values than the standard *L. leichmannii* method.

The extraction process in the *L. leichmannii* assay has now been used without modification for almost 20 years. Undoubtedly the reason for the continued belief that deproteination in the presence of cyanide produces complete extraction of vitamin $B_{12}$ has been the use of the recovery experiment. Recovery by the *L. leichmannii* assay of vitamin $B_{12}$ added to serum has been excellent even since cyanide was introduced into the extraction process, and the recovery results shown in this paper are typical of others' experience. The usual assumption has been that if any vitamin $B_{12}$ assay, microbiological or radio-isotopic, can recover 100% of (exogenous) vitamin $B_{12}$ added to a serum, then the extraction of the endogenous vitamin $B_{12}$ in the serum is also complete (Jacobs and Zondag, 1969; Frenkel *et al*, 1970). It is most unlikely that this assumption is correct. Endogenous vitamin $B_{12}$ in serum is tightly bound largely to transcobalamin I (Benson *et al*, 1972); vitamin $B_{12}$ added to serum is bound in a labile fashion largely to transcobalamin 2 from which vitamin $B_{12}$ can be separated by simple measures such as heating at 56°C for 30 minutes (Gullberg, 1970). Accordingly, a more severe extraction process is required for the assay of endogenous vitamin $B_{12}$ than for the recovery of vitamin $B_{12}$ added to serum, and an extraction process which is suitable for the recovery of added vitamin $B_{12}$ is not necessarily efficient for the extraction of endogenous vitamin $B_{12}$.

In general, those radio-isotopic assays which use an acid extraction process, either hydrochloric or glutamic acid, produce higher serum vitamin $B_{12}$ values than the *L. leichmannii* method (Raven *et al*, 1969; van de Wiel *et al*, 1974; Raven and Robson, 1974). Those radio-isotopic assays which use a deproteination extraction process produce values in agreement with the *L. leichmannii* method (Matthews *et al*, 1967; Britt *et al*, 1969; Wagstaff and Broughton, 1971). The only exception is the radio-isotopic method of Green *et al* (1974), which uses a deproteination extraction process yet has the highest normal range of all the radio-isotopic assays described so far (400-1100 pg/ml).

In two other radio-isotopic assays (Rothenberg, 1968; Jacobs and Zondag, 1969) an acid extraction process has been found to produce higher serum vitamin $B_{12}$ values than a deproteination method. Rothenberg suggested that the higher values obtained for pernicious anaemia sera with an acid extraction process were falsely high due to interference by denatured proteins with the vitamin $B_{12}$ binder (transcobalamin I). He thought that the lower values obtained with the deproteination extraction process could not be due to less efficient extraction since recovery of added $^{57}$Co$B_{12}$ by the deproteination method was of the order of 95%. Apart from the confusion of extraction and recovery, mentioned previously, this cannot provide an explanation for Rothenberg's results or for the radio-isotopic method of Raven *et al* (1969) producing higher serum vitamin $B_{12}$ values than the *L. leichmannii* method. It is true that denatured serum proteins affect the binding of vitamin $B_{12}$ by intrinsic factor and so play an important part in radio-isotopic assays which use intrinsic factor as the vitamin $B_{12}$ binder (Rothenberg, 1963; Raven *et al*, 1968; Hillman *et al*, 1969; Raven, 1970; Mortensen, 1972). Similar findings have been reported for a radio-isotopic method using an acid extraction process and serum as the vitamin $B_{12}$ binder (van de Wiel *et al*, 1974) but the effect of the denatured proteins is to increase the binding of vitamin $B_{12}$ by intrinsic factor or the serum protein binder—Mortensen (1972) used the term 'negative interference'. In a radio-isotopic assay such as that of Rothenberg (1968) in which there are no serum proteins in the standards, the presence of the denatured proteins produced by an acid extraction process would lead to falsely low serum vitamin $B_{12}$ values if it were not outweighed by improved extraction of vitamin $B_{12}$ from the serum proteins. Rothenberg's argument, however, is an
important one since it is likely to be repeated as it becomes apparent that an acid extraction process is associated with increased serum vitamin B₁₂ values. It would be simple to explain the high values obtained by the method of Raven et al (1969) as being due to the presence of some interfering protein degradation product. One would then expect to obtain lower values when a deproteinized serum extract was used and also to detect the same hypothetical degradation protein in the protein precipitates formed during the *L. leichmannii* extraction process. This argument was the main stimulus for the attempts, described in this paper, to modify the *L. leichmannii* method so that higher serum vitamin B₁₂ values could be obtained. Since it did prove possible by both acid and papain modifications to obtain higher results by the *L. leichmannii* method, it is unnecessary to postulate the presence of some serum degradation product causing artefactually high serum vitamin B₁₂ values by the method of Raven et al (1969).

All the studies in this paper refer to the extraction process used by the *L. leichmannii* method of Chanarin (1969). Is it possible to apply the results to the other varieties of *L. leichmannii* assay in common use? The extraction process of Chanarin (1969) is almost the same as that of Dacie and Lewis (1968) but it might be argued that the extraction processes used by the *L. leichmannii* methods of Spray (1955) and Matthews (1962) are more efficient because they both use twice as much acetate buffer per millilitre of serum as the methods of Chanarin (1969) and Dacie and Lewis (1968). However, this does not appear to be so since the reported lower limits of the normal ranges for these two methods are no higher than that of the method of Chanarin (1969) (Spray, 1955: 150 pg/ml; Spray and Witts, 1958: 150 pg/ml; Batata et al, 1967: 140 pg/ml; Matthews, 1962: 120 and 115 pg/ml). The likelihood remains that the radioisotopic method of Raven (1969) is producing generally higher values than different varieties of the *L. leichmannii* method because it is using a more efficient extraction process to separate serum vitamin B₁₂ from its binding proteins.

We should like to thank Mr J. Holliday, FAIMT, for carrying out the *L. leichmannii* assays shown in figure 10.

References


