Cobalmins and nitrous oxide: a review

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The anaesthetic gas, nitrous oxide (N₂O), once regarded as chemically inert, oxidises some forms of vitamin B₁₂. It does this both when used clinically and in the test tube, and the action is remarkably selective. As far as we know, no other pathway or substance is affected except as a result of damage to vitamin B₁₂. Vitamin B₁₂ that has been oxidised in this way no longer functions as a coenzyme. Thus the effect of N₂O presents the biochemist and haematologist with a remarkable tool with which to explore the mode of action of vitamin B₁₂. It presents the neurologist and neuropathologist with a new probe into the mechanism of vitamin B₁₂ neuropathy, and finally it is a new tool with which to explore the complex field of vitamin B₁₂-folate interrelations.

The purpose of this review is to discuss current work in these fields. No attempt is made to discuss N₂O from the viewpoint of an anaesthetic agent, and the review deals only with the effect on vitamin B₁₂.

Chemistry

In vitro a relationship between vitamin B₁₂ and N₂O was demonstrated by Banks et al.¹ and others.² Vitamin B₁₂ is one of a group of transition-metal complexes having a metallic ion linked directly to an organic compound. These are able to activate N₂O, releasing free nitrogen and oxygen. The B₁₂ itself is changed rapidly from the reduced cob(I)alamin or B₁₂ form to the oxidised cob(III)alamin or B₁₂ form, a change accompanied by rapid conversion of the grey-green colour of cob(I)alamin to a reddish-brown colour due to a mixture of cob(II) and cob(III)alamin. The latter is inactive in such pathways as methionine synthetase, which require cobalt in the fully reduced state.

\[
\text{Cob(I)alamin} + \text{N}_2\text{O} \rightarrow \text{Cob(III)alamin} + \text{H}_2\text{O} + \text{N}_2
\]

Cob(III)alamin + H₂O + N₂ → 2 cob(II)alamin

In the intact animal it is likely that the enzyme as well as the coenzyme (B₁₂) is damaged, presumably secondary to damage to the B₁₂, because recovery from N₂O exposure takes several days, and recovery is not more rapid if B₁₂ is supplied, suggesting that new apoenzyme needs to be synthesised.

Clinical observations

The first clinical report of toxicity that could be ascribed to N₂O was that of Lassen et al. in 1956,⁶ who used a 50% N₂O/oxygen mixture as well as other agents to control the spasms in patients with tetanus. Treatment was continued for up to six days. Two of the patients died, and pancytopenias appeared in most of the patients accompanied by megaloblastic haemopoiesis demonstrated by marrow aspiration. The marrow, indeed, was indistinguishable from that in untreated pernicious anaemia. A similar experience was reported in a further case from Australia.⁴ Amess et al.,⁵ in studying the effects of N₂O during and after open heart surgery, found that the marrow was megaloblastic and, by use of the deoxyuridine suppression test, they showed that the pattern of behaviour was similar to that of marrow in untreated pernicious anaemia where there is some return towards normality in the test by the addition of vitamin B₁₂. There was a greater improvement on the addition of folate.

At about the same time reports of a neuropathy in those exposed to excess amounts of N₂O began to appear in the medical literature. By far the largest group were dentists who had developed some addiction to N₂O inhalation, but a dentist and his assistant, who were exposed to N₂O from a leak in defective anaesthetic equipment, were both affected, as was a young lady who obtained her N₂O from capsules designed to produce whipped cream in the home.⁶⁻⁹

The commonest early symptom was numbness and tingling in the hands or feet. A few noticed loss of dexterity of the fingers, poor balance, or leg weakness. One patient first noticed continuous twitching of two toes on one foot. At first the numbness tended to be patchy but symmetrical. Eventually numbness of the legs was present in all the patients, and in most the hands were also affected. Many became unsteady because of sensory loss and about half were unable to walk unassisted. Neck flexion produced a shock sensation radiating down to the legs or, in one case,
the arms (Lhermitte sign). Impotence, difficulty with micturition, and constipation were common. Half the patients were depressed with impaired memory and had difficulty in thinking clearly. Ten out of 15 patients had to stop work as a result of their neuropathy.6

Examination in the early stages showed only reduced knee and ankle jerks and slight loss of touch and vibration sense in the extremities. Subsequently, the tendon reflexes became brisk with a positive Babinski sign. This was accompanied by severe loss of sensation affecting limbs and trunk, positive Romberg sign, and a wide-based ataxic gait.

Electromyography showed denervation, and motor nerve conduction tests reduced conduction velocity in all patients tested. Sensory nerve testing showed a reduced amplitude or increased latency of evoked potentials in five out of six cases.

Examination of cerebrospinal fluid was generally normal (one patient had increased protein). The serum vitamin B12 level was normal in seven cases but was 170 and 185 pg/ml in two others. Absorption of vitamin B12 was normal. One patient had hypersegmented neutrophils in the blood film.

A large survey has been carried out in the USA among dentists who use N2O for anaesthesia as compared to those who do not. In male dentists using N2O there was a 75% increase in liver disease, a 70% increase in neurological disease, and a 55% increase in spontaneous abortion in their wives as compared to the control group.10 Among female dental assistants exposed to N2O there was, in addition, a 155% increase in the abortion rate, an 80% increase in congenital malformations, and even a 90% increase in cancer as compared to controls.

These reports have stimulated considerable interest and even some activity. An understanding of the effect of N2O may be important for several reasons:

1 N2O, if it could be used safely, is a valuable and potent anaesthetic, particularly in severely injured patients, after major surgery, and in child birth. Is there some way in which the toxic effects of prolonged use can be prevented?

2 What are the consequences of exposure to low levels of N2O over relatively long periods of time, as may occur among staff working in operating theatres in both general surgery and dentistry?

3 What are the biochemical consequences of B12 inactivation in man and the experimental animal, and does it offer a model in which to study the effects of B12 deficiency, including the changes in the nervous system and the effects on folate metabolism?

4 Are the analgesic and para-anaesthetic effects influenced by the same or different pathways from those that affect B12?

General remarks

Vitamin B12 functions as a coenzyme. Ten enzymatic reactions have been identified requiring adenosylcobalamin but only one, methylmalonyl-CoA mutase, occurs in mammals including man.11 Several reactions require a second B12 analogue, methylcobalamin, but only one of these pathways, methionine synthetase, has been found in mammals. Either independently of these reactions, or through their agency, vitamin B12 is necessary for the normal functioning of the folate coenzymes. Thus N2O affects not only the B12 coenzymes but also the folate coenzymes.

Diseases due to B12 deficiency occur in man and in ruminants grazed on cobalt-deficient pastures. Neither has proved to be an ideal model for study, man because therapy rapidly reverses the picture and because, for ethical reasons, tissues such as liver are not normally available, and ruminants because their size makes them too awkward to manage, although valuable studies have been carried out on sheep in Australia.12 13 Dietary depletion can produce partial B12 deficiency in rats, and a clean fruit diet leads to B12 neuropathy in fruit bats.14 Man alone, however, develops a megaloblastic anaemia after B12 deficiency, and ultimately we will have to look to human data to resolve this problem.

At the time of writing, the effects of N2O are being studied to a variable extent in man, monkey, rat, mouse, fruit bat, and some bacteria.

EFFECTS OF N2O ON VITAMIN B12 MEDIATED PATHWAYS

Methionine synthetase

In this reaction homocysteine takes up a methyl group to form methionine. The source of the methyl is usually either the β-carbon of serine or the α-carbon of glycine. The carbon unit is taken up by folate to form 5,10-methylenetetrahydrofolate, which then transfers to cob(II)alamin to give methylenetetrahydrofolate. The methyl group is then transferred to cob(III)alamin to give methylcobalamin, and this is the final methyl donor.

Direct assay of enzyme activity in the liver of Sprague-Dawley rats exposed to a 50% N2O/oxygen mixture showed a substantial fall after 30 minutes' exposure, and after 6 hours methionine synthetase activity was at a virtually zero level.15 16 The levels remain virtually undetectable for as long as exposure to N2O is continued, this particular study being terminated after 15 days. Recovery of activity after withdrawal of N2O is relatively slow, there being a steady restoration over not less than four days.
Brain methionine synthetase is depressed to an equal degree, and this too persists throughout the period of exposure to N₂O.¹⁷ There is no accompanying homocysteinuria. Kondo et al.¹⁸ preceded exposure to N₂O with an injection of ^⁵⁷⁷Co-B₁₂. Chromatography was used to separate labelled B₁₂ on the hepatic mutase and synthetase. With N₂O exposure there was a steady fall in labelled B₁₂ in the position of methionine synthetase.

The long recovery period after withdrawal of N₂O suggests that the apoenzyme is degraded after loss of B₁₂, and the delay in recovery is due to synthesis of new apoenzymes.

**Methylmalonyl-CoA-mutase**

The conversion of methylmalonic acid (MMA) to succinic acid requires adenosylcobalamin as coenzyme. MMA arises from metabolism of propionic acid, valine, leucine, and isoleucine. It may be excreted in the urine in B₁₂ deficiency, particularly if a precursor substance has been given. MMA was not detected other than in trace amounts in the urine of rats exposed to N₂O even after an injection of propionic acid, whereas a group of rats on a B₁₂-deficient diet increased their excretion from 0·1 mg to 36 mg in 24 hours.¹⁵ It was concluded that the form of B₁₂ in this reaction was not susceptible to the oxidative action of N₂O. This could be because the reaction did not require a reduced cobalamin. Although methionine synthetase is a cytosol enzyme and the mutase a mitochondrial one, this is probably not a likely explanation for the difference in behaviour as N₂O readily penetrates cell organelles such as mitochondria. Kondo et al.¹⁸ confirmed that in liver from an animal given ^⁵⁷⁷Co-B₁₂ the radioactivity on the mutase remained intact, unlike that on the synthetase. A report that there is an increased excretion of MMA in mice receiving 80% N₂O has been withdrawn.¹⁹

**Vitamin B₁₂ coenzyme levels**

The loss of methylcobalamin in rat liver and the maintenance of B₁₂ on the mutase, presumably adenosylcobalamin, demonstrated by Kondo et al.¹⁸ has been mentioned. Patients developing neuropathy due to N₂O inhalation generally have normal serum B₁₂ levels.⁷ Linnell et al.²⁰ exposed human lymphocytes, which had been incubated with phytohaemagglutinin and ^⁵⁷⁷Co-B₁₂ for 72 hours, to 60 seconds’ N₂O. This was followed by a decline in the amount of methylcobalamin in the cell while the level of adenosylcobalamin was maintained.

**Ethanolamine ammonia-lyase**

This clostridial enzyme converts ethanolamine to acetaldehyde. It too is inactivated by exposure to N₂O, implying a reduced cobalt derivative in the adenosylcobalamin.²¹ It does not occur in mammals.

### Effects of N₂O on Folate and Folate Coenzymes

**Hepatic uptake of injected folate**

Using mice, McGing et al.²² showed that ^³H-labelled pteroylglutamic acid (^³H-PteGlu), given intraperitoneally, was taken up poorly by liver after N₂O inhalation. The uptake after N₂O was reduced to only 40% of the amount taken up by controls. The hepatic uptake of physiological folate analogues is also impaired after N₂O inhalation.²³ The results in Table 1 show that 21% of tetrahydropteroylglutamic acid (H₄PteGlu) was taken up by liver as compared to the uptake in a control animal. The normal transport form of folate is 5-methyltetrahydrofollic acid (5-CH₃-H₄PteGlu) and, following N₂O, its hepatic uptake was reduced to 38% of the uptake in an untreated animal. The formylfollates are less affected, values ranging from 56 to 69%.²⁴ of that in controls.

| Table 1 Uptake of labelled folate analogues, given intraperitoneally to rats, by liver and their conversion into folatepolyglutamate²³ |
|---------------------------------|------------------|------------------|
| Folate | IP folate in liver (%) | IP folate as polyglutamate (%) |
|        | Control | N₂O-treated | Control | Treated |
| H₄PteGlu | 3·7 | 0·78 (21) | 55 | 0 |
| 5CH₃H₄PteGlu | 4·2 | 1·6 (38) | 42 | 0 |
| 5CHOH₂H₄PteGlu | 5·3 | 2·96 (56) | 52 | 49 |
| 10CHOH₂H₄PteGlu | 6·7 | 4·63 (69) | 52 | 46 |
| 5,10CH₂H₂H₄PteGlu | 5·06 | 3·47 (69) | 55 | 59 |

**Synthesis of folatepolyglutamate**

The active form of folate functioning as a coenzyme is a polyglutamate, that is, it has additional glutamic acid residues, usually four, to form a pentaglutamate. The glutamic acid peptide chain serves to attach the coenzyme to the apoenzyme. The enzyme adding on glutamic acid units, folatepolyglutamate synthetase or ligase, is present in all cells. After N₂O, McGing et al.²² found a decline in polyglutamate formation in the livers of mice given the oxidised PteGlu, and the results in rats, again with physiological analogues, are shown in Table 1. There was no detectable polyglutamate formation with either H₄PteGlu or 5-CH₃-H₄PteGlu as substrates. However, folate polyglutamate formation was entirely normal when formyltetrahydrofolic acid (CHO-H₄PteGlu) was given.

The data indicate that polyglutamate formation itself is unaffected by inactivation of B₁₂, and the
results imply that B₁₂ is concerned in the provision of the correct substrate for polyglutamate synthesis. The data further imply that the correct substrate is formyltetrahydrofolate. By implication B₁₂ is concerned in this formulation step.

*Serum folate levels*

The serum folate level is usually normal in untreated pernicious anaemia, but in some patients the serum level is increased. After N₂O inhalation in rats Lumb *et al.* (unpublished observations) found a very marked increase in serum folate level within several hours of exposure, and this level remained elevated while N₂O inhalation was continued. 5-Methyltetrahydrofolate undergoes an enterohepatic circulation, and the rise in serum methylfolate level appears to be due to interruption of the enterohepatic cycle by the impaired uptake of the methylfolate by liver (see Table 1).

*Liver folate*

The liver is a major site of folate metabolism and holds half or more of the folate in the body. The amount and form of folate in liver following N₂O was of interest because it provided an opportunity for testing the 'methylfolate trap hypothesis'. This hypothesis states that failure of the methionine synthetase reaction, in which the methyl group of methylfolate is transferred to homocysteine, leads to an accumulation of methylfolate and hence to lack of other folate analogues. Elevation of serum folate may occur in untreated pernicious anaemia and favours the concept of accumulation of methylfolate but the red cell folate is reduced in two-thirds of patients so that there is no pile-up of methylfolate at this site. It has proved ethically impossible to obtain pretreatment tissues from other sites, such as liver biopsy material, in untreated pernicious anaemia to test the hypothesis further.

Inactivation of the methionine synthetase path by N₂O offered an opportunity to determine if, in fact, methylfolate was trapped or whether it was metabolised by means other than through the homocysteine-methionine reaction.

Figure 1 shows that rats maintained in an atmosphere of 50% N₂O/50% oxygen for 10 days lose folate continuously from the liver. This loss amounts to 50% of total folate in the first 48 hours and to 75% of folate after 10 days. The rate of loss is greatest with folates supporting the growth of *Lactobacillus casei* (methyltetrahydrofolate as well as other monoglutamate analogues) and is less rapid with *Pediococcus cerevisiae*-active folates (reduced folate analogues other than methylfolate).

More detailed analysis of methylfolates over the first three days (Fig. 2) shows that there is a rise in 5-methyltetrahydropteroylpolyglutamates at 24 hours but no rise of monoglutamates. This rise, however, disappears at 48 hours. The accumulation of methylpolyglutamate is due to cessation of methionine synthetase activity. There is no 'trapping' at the monoglutamate level as postulated in the hypothesis and hence the reaction cannot be concerned with provision of the substrate for folate polyglutamate synthesis. Secondly, the trapping, which not unexpectedly occurs at the polyglutamate or active...
coenzyme level, is transient. At 48 hours the methylpolyglutamate has been metabolised through other pathways, one of which may be the oxidation of the methyl group back to methylene via methylene-tetrahydrofolic reductase, a reaction which has been demonstrated in vitro in relation to methylation of biogenic amines.27-29
This spectacular fall in hepatic folate is in part due to interruption of the enterohepatic circulation of methyltetrahydrofolic. Liver methylfolate is excreted into bile and thence into small gut where it is reabsorbed and again taken up by the liver. The hepatic uptake of methylfolate after an intraperitoneal injection in an N2O-treated rat is reduced to one-third of that in control animals. The second factor leading to a fall in hepatic folate after exposure to N2O is a virtual cessation of polyglutamate synthesis from methylfolate.

**Folate in other tissues**
The change in folate levels in kidney, brain, and marrow after exposure to N2O follow the pattern of liver, although the fall in total folate levels is not as great.

**Folate catabolism**
The massive loss of folate after N2O implies excretion and/or catabolism of folate, but as yet no data have been published.

**Utilisation of deoxyuridine by bone marrow**
The utilisation of deoxyuridine for thymidine synthesis, a folate-dependent step, is tested in the deoxyuridine suppression test using bone marrow.30 31 Normal human or rat bone marrow cells will meet more than 90% of their thymidine requirements by methylation of deoxyuridine (synthetic path), and less than 10% of 3H-thymidine, added to the marrow cell suspension at a later stage, appears in DNA (salvage path). In megaloblastic anaemia there is reduced synthesis of thymidine for DNA synthesis. Thus the uptake of 3H-thymidine in megaloblastic anaemia exceeds 10%.

Marrows from patients treated with N2O give an abnormal deoxyuridine suppression test,5 and in the study of Amess et al., there was a return towards normality on the addition of vitamin B12 to the test system. Scott et al.19 reported that bone marrow from rats treated with N2O developed an abnormal deoxyuridine suppression, which, too, was improved by addition of B12, PteGlu, and 5-CHOH2PteGlu but not by the addition of 5-CH3H4PteGlu.

A more detailed examination of the role of folate analogues in thymidine synthesis in the N2O-treated rat was carried out by Deacon et al.16 The results (Table 2) were assessed against matched controls, that is, the tubes with 3H-thymidine but without deoxyuridine (= 100% value in the test) also contained whatever other additives had been added to the test mixture. Table 2 shows that the addition of B12 produced no significant improvement, unlike the situation in man and monkey,32 and the greatest improvement followed the addition of formyltetrahydrofolic. H4PteGlu produced relatively poor improvement in the methylation of deoxyuridine and methylfolate, and PteGlu no improvement. Thus those folate analogues that overcame the block in folate polyglutamate synthesis in the N2O-treated rat were also the most effective in restoring a more normal deoxyuridine utilisation.

**Table 2 Results of the deoxyuridine suppression test with marrow cells in control and N2O-treated rats and the effect of addition of hydroxocobalamin and/or folate analogues**

<table>
<thead>
<tr>
<th>Additive</th>
<th>Controls</th>
<th>N2O-treated</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>No. Mean SD</td>
<td>No. Mean SD</td>
</tr>
<tr>
<td>Nil</td>
<td>40 7.3 1.57</td>
<td>21 15.7 2.46</td>
</tr>
<tr>
<td>OH-Cbl</td>
<td>8 6.8 1.62</td>
<td>8 15.0 2.31</td>
</tr>
<tr>
<td>PteGlu</td>
<td>4 6.7 1.17</td>
<td>4 14.7 2.13</td>
</tr>
<tr>
<td>H4PteGlu</td>
<td>8 6.5 0.89</td>
<td>8 12.8 2.02</td>
</tr>
<tr>
<td>5-CH3H4PteGlu</td>
<td>8 6.3 1.55</td>
<td>8 14.5 2.86</td>
</tr>
<tr>
<td>5-CHOH2PteGlu</td>
<td>10 7.5 1.66</td>
<td>10 9.2 1.44</td>
</tr>
<tr>
<td>10-CHOH2PteGlu</td>
<td>8 6.2 1.22</td>
<td>8 9.4 1.89</td>
</tr>
<tr>
<td>OH-Cbl + PteGlu</td>
<td>4 7.2 1.89</td>
<td>4 13.8 1.55</td>
</tr>
<tr>
<td>OH-Cbl + H4PteGlu</td>
<td>8 7.4 1.33</td>
<td>8 13.5 2.90</td>
</tr>
<tr>
<td>OH-Cbl + 5-CH3H4PteGlu</td>
<td>8 6.6 1.57</td>
<td>8 14.1 1.56</td>
</tr>
<tr>
<td>OH-Cbl + 5-CHOH2PteGlu</td>
<td>8 7.5 1.80</td>
<td>8 8.9 1.83</td>
</tr>
<tr>
<td>OH-Cbl + 10-CHOH2PteGlu</td>
<td>8 7.4 1.26</td>
<td>8 12.6 2.58</td>
</tr>
</tbody>
</table>

Results are expressed as: counts with deoxyuridine + [3H]thymidine × 100 counts with [3H]thymidine

**Utilisation of formate**
Lymphocytes from patients with untreated pernicious anaemia fail to synthesise serine from formate and glycine in a normal manner. These are folate-dependent reactions. This was improved only after the patient had been treated with vitamin B12.33 34 Preliminary results with marrow from N2O-treated rats showed a marked impairment of this pathway.

**VITAMIN B12-FOLATE INTERRELATIONS**
In a review of intracellular folate metabolism given at the Annual Meeting of the American Society for Haematology in San Diego, Bertino35 stated that the methylfolate trap was no longer hypothesis, it was fact. As the few new facts that had emerged over the preceding decade had merely served to cast further uncertainty around the theory, this was a surprising statement. The absence of a satisfactory model for B12 deficiency has made it difficult to explore
methylfolate trap hypothesis. The effect of N₂O, firstly, as a means of inactivating methionine synthesis and, secondly, blocking the path by which B₁₂ regulates folate metabolism, has provided the first convenient opportunity for testing this hypothesis in depth.

N₂O in the rat causes a transient accumulation of 5-methyltetrahydropteroylpolyglutamate, and thereafter either other pathways are induced, which brings about further metabolism of methylfolate, or its synthesis is shut off. The latter could occur if S-adenosylmethionine, which produces feedback inhibition of methylenetetrahydrofolate reductase, were shown to accumulate. This seems unlikely as synthesis of methionine virtually ceases. After this initial stage there is no further evidence of trapping of methylfolate in any tissue. There is a steady disappearance of folate from liver, rather than an accumulation in any form, and this indicates that there is no methylfolate trap.

The effect of N₂O is bypassed, as far as folatepolyglutamate synthesis and deoxyuridine methylation are concerned, by the provision of formyltetrahydrofolate derivatives. Other folate analogues, and particularly unsubstituted tetrahydrofolate, were far less effective or not effective at all. Thus the role of B₁₂ appears to be in the formylation of tetrahydrofolate. The source of formate is serine and/or glycine, and the formation of serine from ¹⁴C-formate and glycine was found to be impaired in all 16 patients with B₁₂-deficient megaloblastic anaemia studied by Tikerpae and Chanarin. Further elucidation of the role of B₁₂ is awaited.

NEUROLOGICAL EFFECTS OF N₂O

The clinical observations indicating that exposure to N₂O produces a neuropathy in man have been mentioned earlier in this review. Dinn et al. failed to produce neurological changes in rats after eight months’ exposure to 50% N₂O. However, they found that a monkey, after two months in an N₂O atmosphere, became unsteady and uncoordinated with progressive ataxia, and finally the animal was unable to sit up or drink. Histological examination of the spinal cord showed degeneration of both the myelin sheath and axis cylinders in the posterior columns as well as in the lateral corticospinal and spinocerebellar tracts. In several areas of the posterior columns, entire fibre tracts were replaced by groups of fatty macrophages giving an appearance of spongy degeneration affecting also the anterior columns. The central grey matter and myelinated tracts bordering it were intact.

Biochemically methionine synthetase activity in rat brain was virtually undetectable 24 hours after 50% N₂O. It was not measured earlier than after 24 hours’ exposure. There was a transient rise in the level of 5-methyltetrahydropteroylpolyglutamate as in liver, which then declined, and a fall in total folate, the decline being at a slower rate than in liver (Lumb et al., in preparation).

At the time of writing, the changes in brain are the same as the changes in other tissues, and as yet there is no evidence that B₁₂ neuropathy is based on interruption of different pathways from that in liver.

HAEMATOLOGICAL EFFECTS OF N₂O

It should be remembered that while folate deficiency causes megaloblastic anaemia in all species, vitamin B₁₂ deficiency causes megaloblastic anaemia only in man. Nevertheless the biochemical sequelae of B₁₂ lack are very similar in many species, and these include changes in the marrow of the sort measured in the deoxyuridine suppression test.

Severe neutropenia and thrombocytopenia followed the administration of N₂O to man over several days, and in the study of Lassen et al., as in that of Amess et al., it is clear that haemopoiesis had become megaloblastic. In the study of Amess et al. all the patients receiving N₂O for 24 hours had megaloblastic marrows. Another group received N₂O only during the operation (elsewhere the time is given as less than 10 hours), and only three out of nine had megaloblastic haemopoiesis. One of the patients with neuropathy was noted to have hypersegmented neutrophils.

Different strains of rat vary in their sensitivity to N₂O. Sprague-Dawley rats develop leucopenia within two to three days of exposure to 50% N₂O. Male LEW/1, Mai rats develop marrow hypocellularity after three days on 40% N₂O but only after 14-21 days with 20% N₂O. There is vacuolation of myeloid precursors. Lymphopenia was an early feature. Recovery occurred three days after withdrawal of N₂O. After about 10 days tolerance to the effect of 40% N₂O developed in the animals.

Human marrows collected 24 hours after 50% N₂O showed an increased number of cells in the early synthetic phase (S) of DNA synthesis.

OTHER EFFECTS OF N₂O

It is surprising how well rats adjust to 50% N₂O. They appear to eat normally and be as lively as normal rats. Pope et al. exposed Sprague-Dawley rats to 8 hours 50% N₂O daily for the 21-day period of gestation and did not produce fetal loss significantly different from that of a control group. There was, however, a decrease in average fetal weight (common to N₂O and other anaesthetic gases such as halothane and methoxyflurane), and this was accompanied by a decrease in the numbers of ossification centres in the fetal vertebral columns.
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However, continuous exposure of pregnant rats to N\textsubscript{2}O (70-80\%) for 24 hours on the ninth day of gestation resulted in decreased numbers of fetuses and an increase in fetal malformation.\textsuperscript{42 43}

Conclusion

This survey indicates that N\textsubscript{2}O is able to shut off major segments of B\textsubscript{12} metabolism in both man and animals. Whether all the effects are through inactivation of methionine synthetase is not certain, but this seems unlikely. The nature of other pathways involved is being studied.

References


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